

Al-Ni-Rh (Aluminum-Nickel-Rhodium)

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Very recently, [2007Prz] investigated the phase relationships in the Al-rich region of this ternary system and presented partial isothermal sections at 1080, 1000, 900, and 800 °C. Besides the quasicrystalline decagonal phase D, a ternary phase of hexagonal symmetry labeled χ and another phase labeled x of unknown structure were found.

Binary Systems

The Al-Ni phase diagram [1993Oka] shows five intermediate phases: NiAl₃ ($D0_{11}$, Fe₃C-type orthorhombic), Ni₂Al₃ ($D5_{13}$ -type hexagonal, denoted δ), NiAl ($B2$, CsCl-type cubic, denoted β), Ni₅Al₃ (Ga_3Pt_5 -type orthorhombic), and Ni₃Al ($L1_2$, AuCu₃-type cubic, denoted γ'). The Al-Rh phase diagram [2006Kho] depicts the following intermediate phases: Rh₂Al₉ ($D8_d$, Co₂Al₉-type monoclinic), Rh_{1-x}Al₃ (orthorhombic, denoted O₁ or ϵ_{16}), RhAl₃ (orthorhombic, denoted O₂ or ϵ_6), Rh₂Al₅(c) (space group $Pm\bar{3}$, cubic, denoted C), Rh₂Al₅(h) ($D8_{11}$, Co₂Al₅-type hexagonal, denoted H), Rh₇Al₃ (monoclinic, denoted V) and RhAl ($B2$, CsCl-type cubic). Ni and Rh form a continuous face-centered cubic solid solution.

Ternary Phase Equilibria

The decagonal D phase has a basic periodicity of ~ 0.4 nm. The χ phase forms between Al₇₆Ni₄Rh₂₀ and Al₇₆Ni₁₃Rh₁₁ and has the hexagonal lattice parameters of $a = 1.2229$ nm and $c = 2.7158$ nm at the composition Al₇₆Ni₈Rh₁₆. The phase x of unknown structure forms around Al₇₀Ni₁₁Rh₁₉ [2007Prz].

With starting metals of 99.999% Al, 99.99% Ni, and 99.95% Rh, [2007Prz] induction-melted under Ar atm a number of alloys. The alloys were annealed at 1080-800 °C for 24-3168 h and quenched in water. The phase equilibria were studied with scanning and transmission electron microscopy, x-ray powder diffraction, energy dispersive x-ray spectroscopy, and differential thermal analysis at heating/cooling rates of 10-50 °C per min. For each alloy, the phases identified and their compositions were listed. All the structurally related ϵ phases were clubbed together and labeled ϵ . The isothermal sections for Al-rich alloys constructed by [2007Prz] at 1080, 1000, 900, and 800 °C are redrawn in Fig. 1-4.

At 1080 °C (Fig. 1), the decagonal D phase is stable around the composition Al₇₁Ni₁₈Rh₁₁ and forms tie-lines with ϵ , Ni₂Al₃, and liquid. The x phase is present around Al₇₀Ni₁₁Rh₁₉. Ni₂Al₃ dissolves up to 4 at.% Rh. The ϵ , C, and V phases dissolve up to 12.5, 10.5, and 3 at.% Ni, respectively. The $B2$ phases NiAl and RhAl probably form a

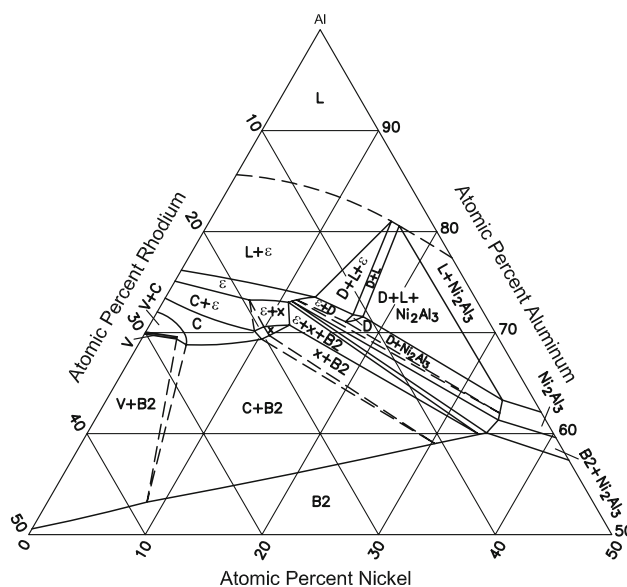


Fig. 1 Al-Ni-Rh isothermal section at 1080 °C for Al-rich alloys [2007Prz]

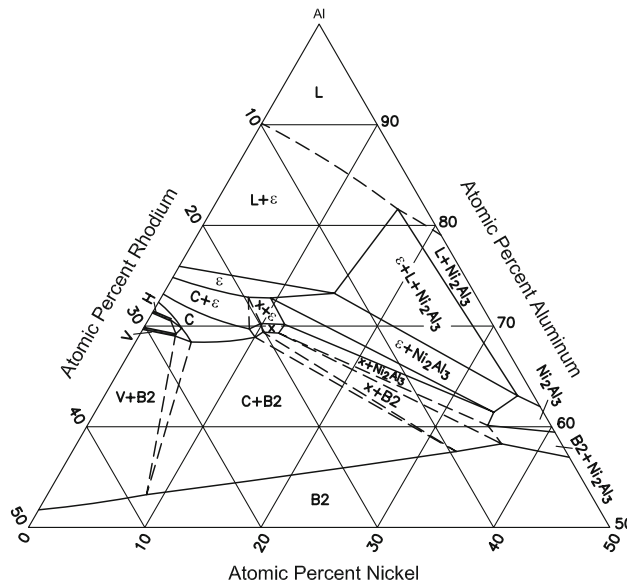


Fig. 2 Al-Ni-Rh isothermal section at 1000 °C for Al-rich alloys [2007Prz]

continuous solid solution (denoted $B2$), which extends up to 60 at.% Al [2007Prz].

At 1000 °C (Fig. 2), the D phase is not stable. The third component solubility in ϵ and Ni₂Al₃ has slightly increased, whereas it remains almost the same in C, V and x phases.

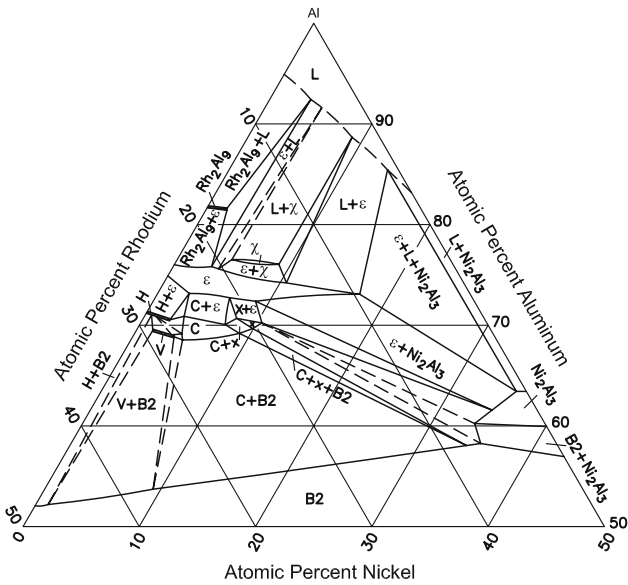


Fig. 3 Al-Ni-Rh isothermal section at 900 °C for Al-rich alloys [2007Prz]

The low-temperature phase H has appeared. At 900 °C (Fig. 3), Rh_2Al_9 is present and dissolves up to 2 at.% Ni. The C and V phases occur only in the ternary region and have ranges of ~ 3 -9 and ~ 1 -3 at.% Ni, respectively. The χ phase is present and has a Ni range of 5-9 at.%. At 800 °C (Fig. 4), the ϵ range is up to 17 at.% Ni. NiAl_3 is present and dissolves up to 3 at.% Rh. The homogeneity range of χ increases to 4-13 at.% Ni. The C phase region shrinks to a narrow strip between $\text{Al}_{69}\text{Ni}_6\text{Rh}_{25}$ and $\text{Al}_{69}\text{Ni}_8\text{Rh}_{23}$. The x phase is not stable at this temperature [2007Prz].

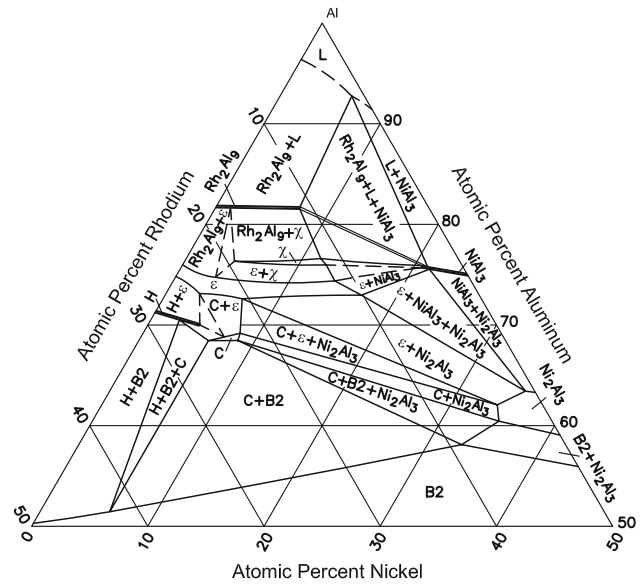


Fig. 4 Al-Ni-Rh isothermal section at 800 °C for Al-rich alloys [2007Prz]

References

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